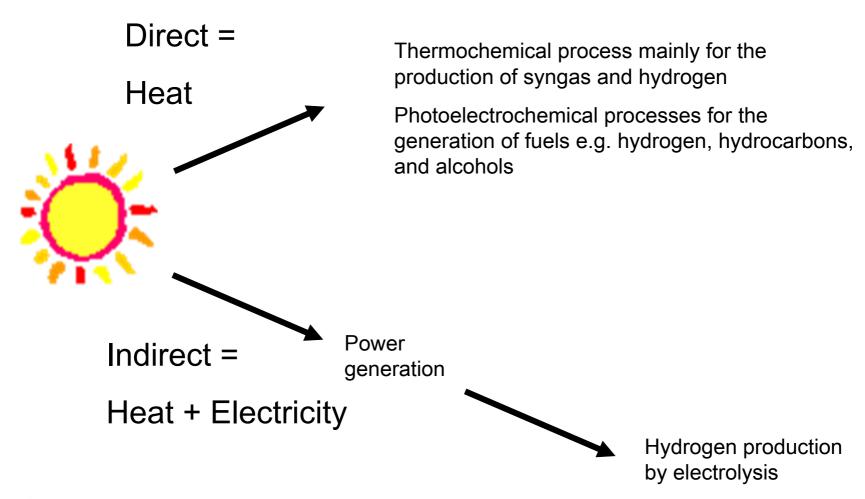


### **Solar Thermal Fuel Production at DLR**

Hans Müller-Steinhagen, Robert Pitz-Paal, Christian Sattler Institute of Technical Thermodynamics German Aerospace Centre (DLR)



#### Conversion of Solar Energy into Solar Fuels



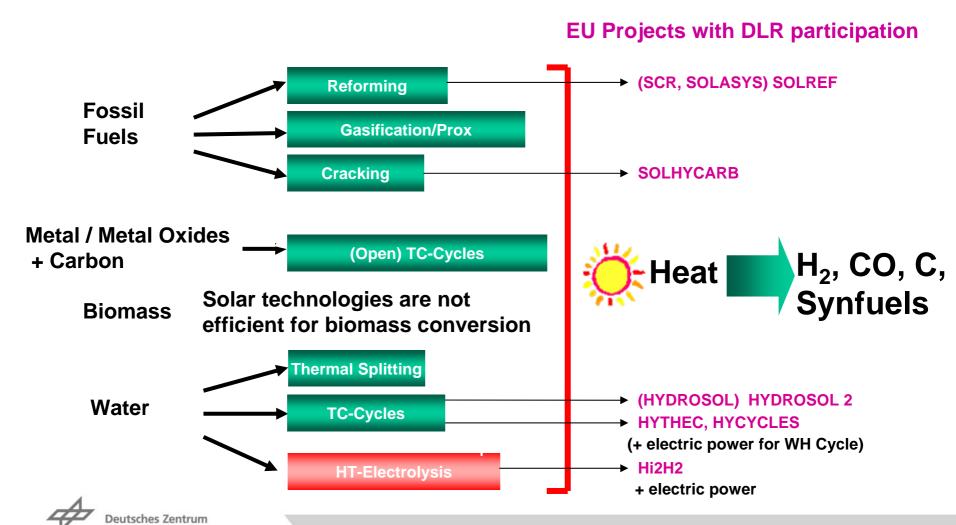
# Criteria for the Selection of Suitable Processes for Solar Thermal Fuel Production

- Tenvironmentally benign and safe process.
- ▼ Technically feasible operation temperature, concentration factor, materials, reaction, throughput.
- High availability of feed-stock.
- **7** Fast reactions with short residence time are preferable.
- → High overall process efficiency.
- Fuel must be produced at acceptable cost

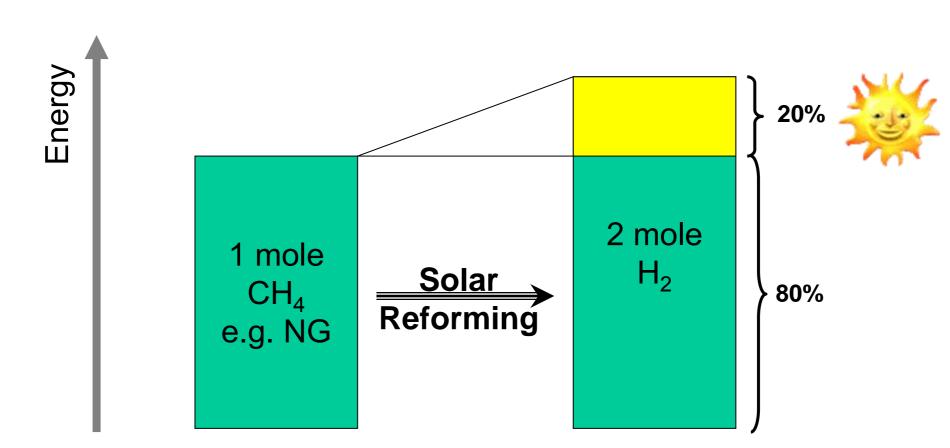


#### **Solar Thermal Fuel Production Routes**

**für Luft- und Raumfahrt** e.V. in der Helmholtz-Gemeinschaft



#### Partly-Solar Hydrogen





#### **Solar Steam Reforming of Natural Gas**

**TEXTERIAL** EXTERIX

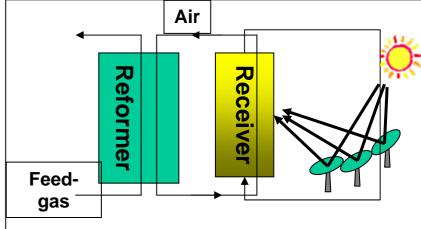
**→ Indirect:** MUSTR, CSIRO

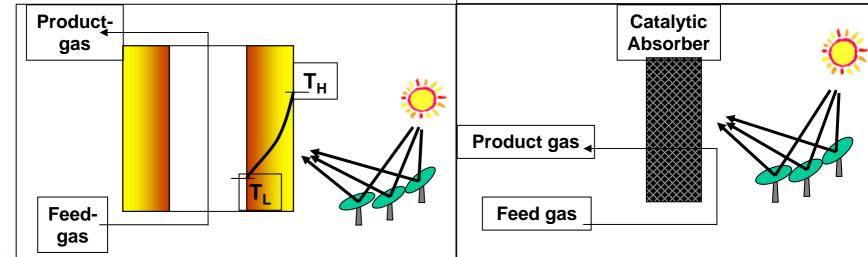
Direct: SCR, SOLASYS, SOLREF

**7** FP 6 Project

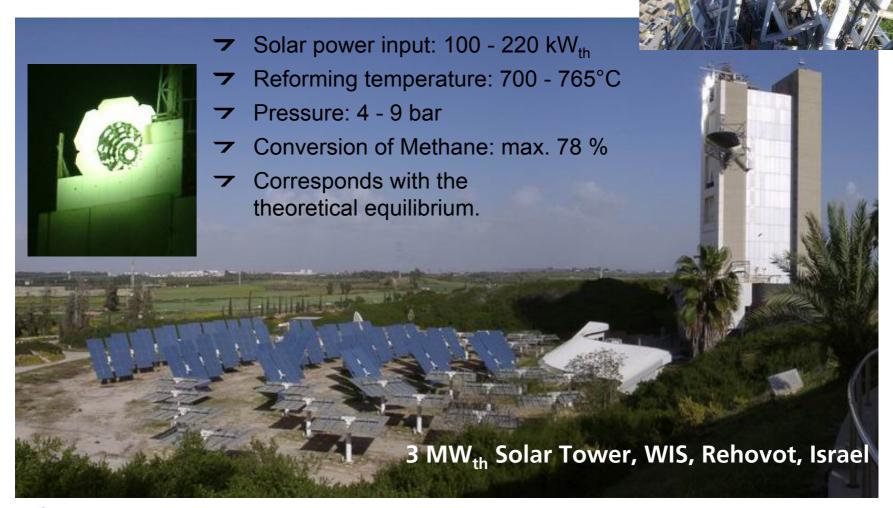
▼ Enhanced Receiver + Demo Model

→ Partner DLR, WIS, ETH/PSI, JM, SHAP, HyGear, CERTH/CPERI





### **Experimental Results SOLASYS (EU FP4)**

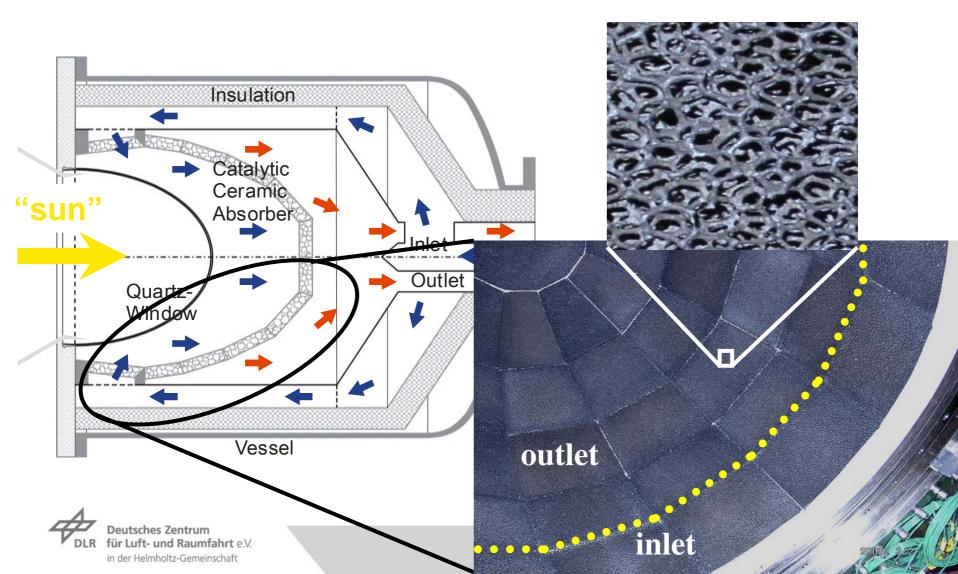


#### **Development of Receiver/Reactor**

- Develop an advanced catalytically-active absorber featuring the following properties:
  - High catalytic activity with high resistance to coking.
  - Good absorption for thermal radiation.
  - Acceptable mechanical strength and thermal shock resistance.
  - → High gas permeability together with high turbulence and mixing of the gases as well as low pressure drop.
  - **7** Low costs.



#### **Schematic of Solar Reformer**

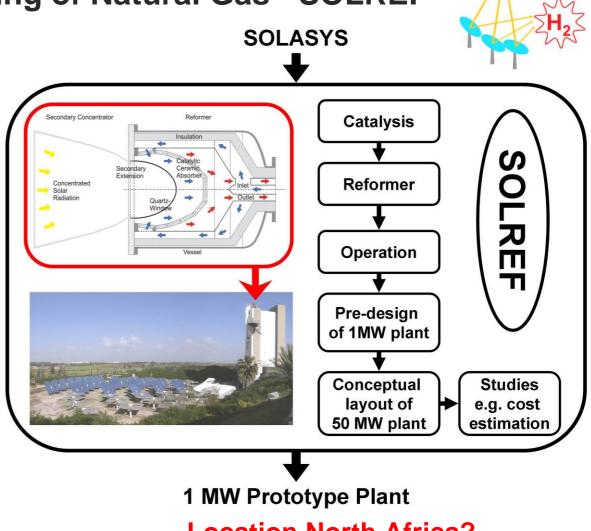


#### Solar Steam Reforming of Natural Gas - SOLREF

2004 - 2008

#### **Consortium:**

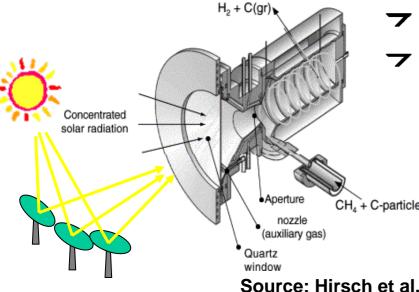
- DLR (DE) (Coordinator)
- CERTH/CPERI (EL)
- WIS (IL)
- ETH (CH)
- Johnson Matthey Fuel Cell Ltd. (UK)
- HyGear B.V. (NL)
- SHAP S.p.A. (I)



**Location North Africa?** 



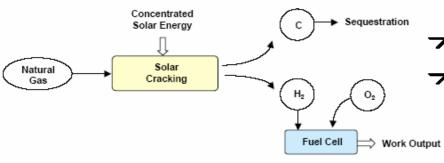
#### H<sub>2</sub>-Production by Solar Cracking of Hydrocarbons



- → Work done by ETH, PSI, CNRS, WIS ...
- Decarbonisation of Methane

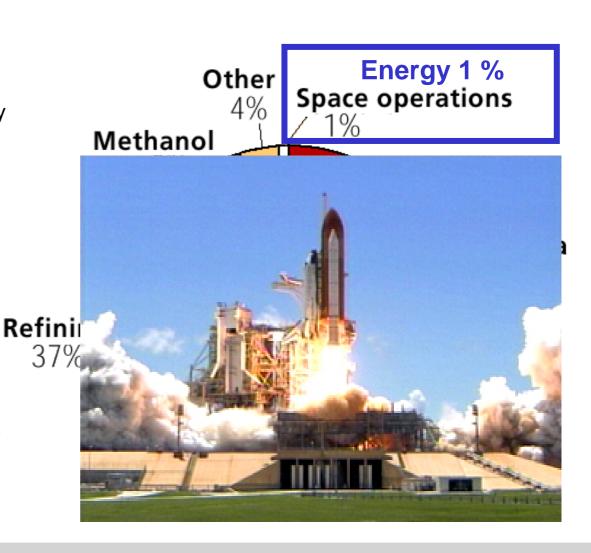
$$CH_4 \leftrightarrow C + 2H_2$$

- **→** Temperature > 1300°C
- → Ambient pressure.
- CH<sub>4</sub> + C-particles Conversion rate 70%.
  - Theoratical system efficiency: 30%
  - Cost: 8 ct/kWh [Dahl et al.] (0-14 ct/kWh depending on the use of the carbon)
  - **→** EU Project SOLHYCARB since March 2006
  - → Partner CNRS/PROMES (FR) Coordinator, ETH, PSI (CH), WIS (IL), CERTH/CPERI (EL), DLR (DE), TIMCAL (BE), SOLUCAR (SP), CREED (FR), N-GHY (FR)



#### Solar, Carbon-Free Hydrogen Production

- ▼ Today, hydrogen is still a bulk chemical rather than an energy vector
- ✓ Annual production:
   600 700 billion Nm³/year
   equal to 53 62 Mt/year
- Virtual value100 billion €/year
- → Growth rate about10 %/year (2003, Linde)
- Only 4 % traded commercially
- → Ammonia production alone generates about 250 Mt CO₂/a





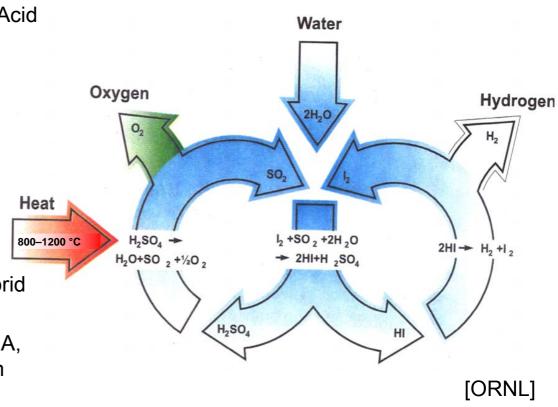
#### Solar, Carbon-Free Hydrogen Production





#### HYTHEC - Sulphur/Iodine and Hybrid Sulfuric Acid Cycle

- Also named General Atomics Cycle or ISPRA Mark 16 and Hybrid Sulfuric Acid Cycle also known as Westinghouse Cycle or ISPRA Mark 11
- EU FP6 STREP HYTHEC
- Solarisation of H<sub>2</sub>SO<sub>4</sub> splitting
- Improvement of processes and increase of efficiency
- Design study for a H<sub>2</sub>/Elctricity Cogeneration plant
- Evaluation of solar, nuclear, and hybrid concepts
- Partners: CEA (coordinator), DLR, EA, Uni Sheffield, Uni Roma Tre, ProSim



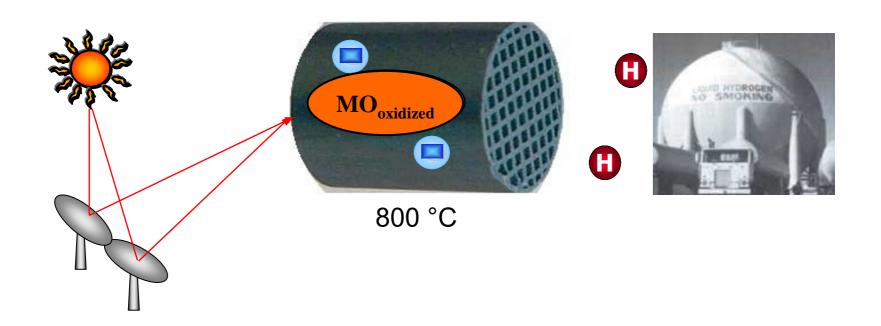


# HYDROSOL: MONOLITH REACTOR FOR HYDROGEN GENERATION FROM SOLAR WATER SPLITTING

- → STREP EU FP 5 (Nov. 2002 Oct. 2005)
- Consortium: CERTH/CPERI APTL (EL, Coordinator), DLR (D), Johnson Matthey Fuel Cell Ltd. (UK), StobbeTech (DK)
- **→** Objectives:
  - → Development of novel active redox materials for the water splitting and regeneration reactions at moderate temperatures (800-1300 °C).
  - → Design, construction, and test operation of a prototype reactor for continuous hydrogen generation based on a thermochemical cycle applying mixed iron oxides
  - Feasibility of operability of solar thermal two-step hydrogen production
  - Evaluation of techno-economic potential of the technology

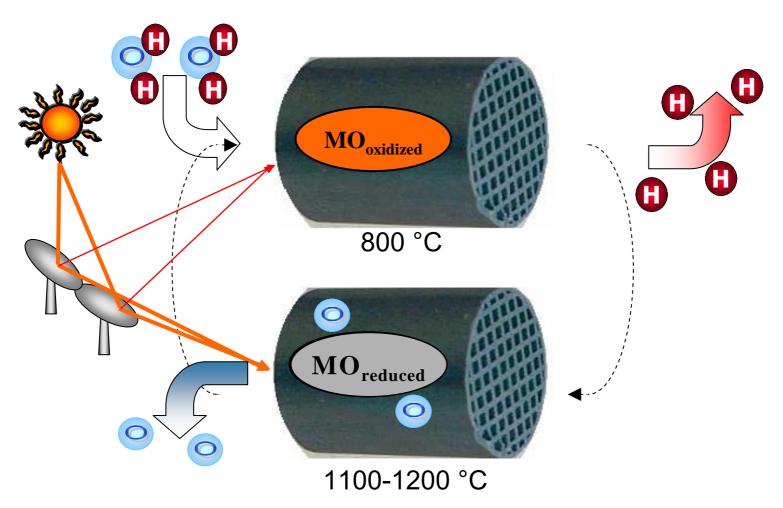


# **HYDROSOL** – Principle of Operation





### **HYDROSOL – Principle of Operation**





# **Batch-Reaktor** after completion



#### **Coated absorber**



#### DLR Solar Furnace, Köln-Porz

Operation started in 1994

Off-axis Concept

Heliostat 60 m<sup>2</sup>

Concentrator 39 m<sup>2</sup>

160 Facettes, 3 Focal lengths

Concentration: 5500 = 5 MW/m<sup>2</sup>

Power max. 25 kW

Focus 13 cm (90%)

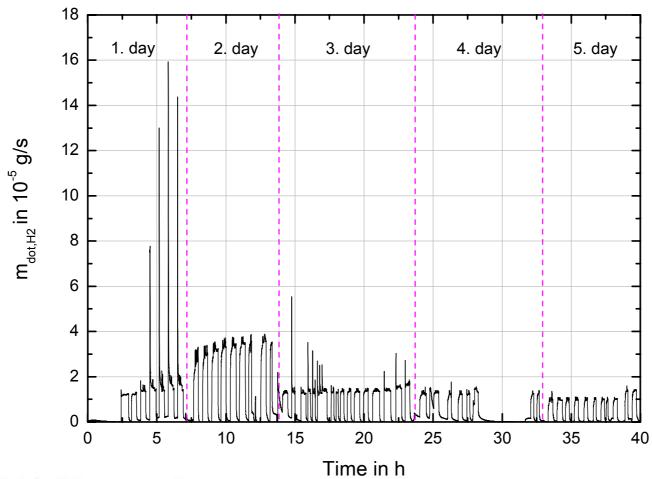
 $T_{max}$  2700 °C





#### Long-term test

Hydrogen generation for 53 cycles performed with one sample





#### Efficiency and characteristics of operation

- ▼ stability of the redox/support-composite proved up to
   40 cycles of constant H₂ production
- quasi-continuously hydrogen generation / alternating reaction conditions
- → SiC and in particular SiSiC turned out as very suitable, robust support material
- Reaction controlled by kinetics

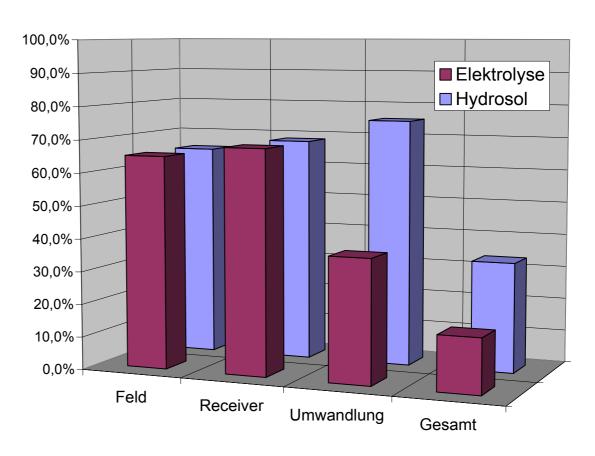
Efficiency of the solar furnace operation:

$$\begin{split} \eta_{\text{ reactor}} &= 0.20 \text{ - } 0.28 \\ \eta_{\text{ (reactor)}} &= \left(Q_{\text{sensible}} + \text{mass flow}_{\text{ (H2)}} * \text{HHV}_{\text{ (H2)}}\right) / Q_{\text{solar}} \\ \eta_{\text{ process}} &= 0.05 \text{ - } 0.09 \\ \eta_{\text{ process}} &= \left(\text{mass flow}_{\text{ (H2)}} * \text{HHV}_{\text{ (H2)}}\right) / Q_{\text{solar}} \end{split}$$



#### Effizienzvergleich HYDROSOL – Elektrolyse

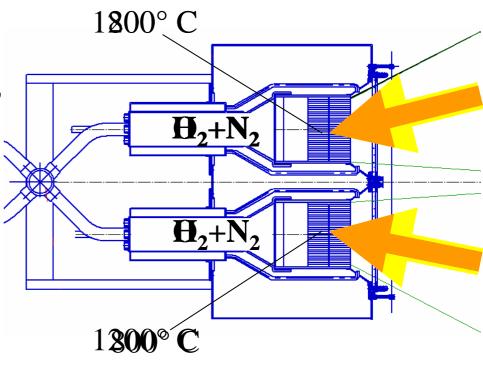






#### Reactor for continuous hydrogen production:

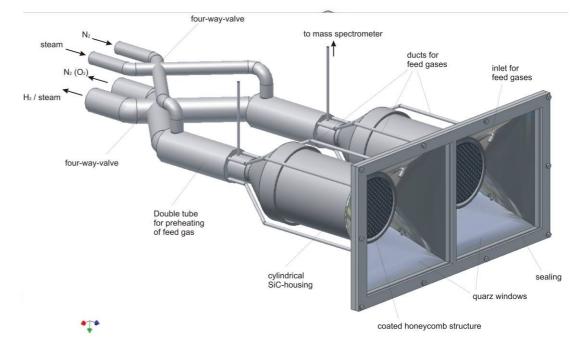
- Reactor with two modules
- Two different alternating processes:
  - Production: 800°C, water steam, nitrogen, exothermic
  - Regeneration: 1200°C, nitrogen, endothermic
- Transient steps like
  - Switching between half cycle
  - Start-up / Shutdown
- Temperature gradient on the coated structure
- Fluctuating irradiation (daily / annually)



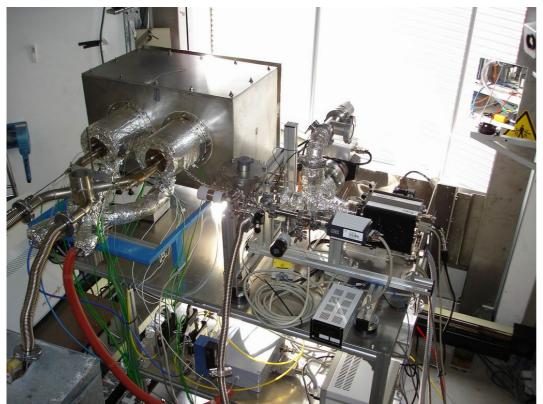


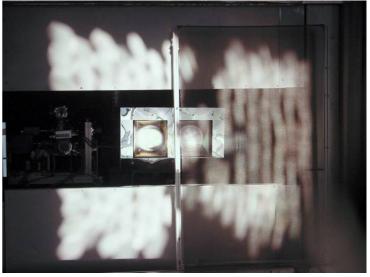
#### Reactor for continuous hydrogen production:

- 15 kW<sub>th</sub> two chamber system
- Continuous hydrogen production
- Four way valve
- Preheating of feed gas
- Measure point for mass-spectrometry
- Quartz window



# The "Conti-Reactor" during test-operation







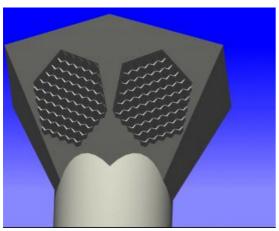
#### **HYDROSOL II**

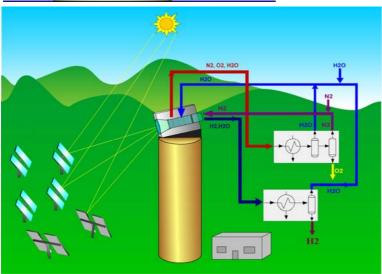
- Start 2006
- New additional Partner CIEMAT
- 100 kW<sub>th</sub>-scale
- Quasi continuous hydrogen production
- 2 modules: production/regeneration
- Use of 9 coated squared ceramic structures (150mm x 150mm x 50mm)
- Installation at SSPS Tower on PSA in the 2nd half of 2007

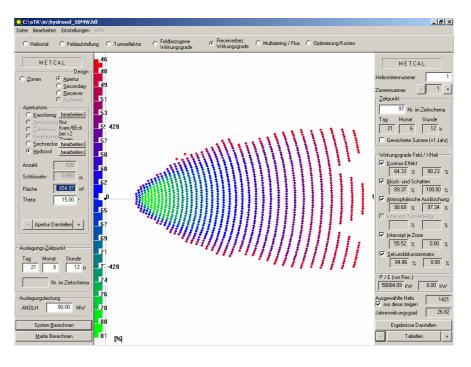




#### **HYDROSOL** – Design Study









#### **Results of Simulation**

- ightharpoonup P<sub>tot,d</sub> = 200 MW at noon on June 21<sup>st</sup>
- → 3264 heliostats needed, which is equivalent to A<sub>mir,tot</sub> = 396 000 m<sup>2</sup>
- **→** Land area of 2.0 km<sup>2</sup>
- → Tower height: 230 m
- Annual production:  $P_{H2}$  = 238 GWh(HHV  $H_2$ )/a or  $V_{H2}$  = 80 Mio. Nm³/a or 7200 t/a.



#### **Investment costs**

Component	Unit price	Units	Total [€]
Heliostats	130 €m²	3264 x 121.34 m <sup>2</sup>	51 500 000
Land	1 M\$/km²	2.0 km <sup>2</sup>	2 000 000
Tower		(230 m)	5 100 000
Buildings	3 M€	1	3 000 000
HT receiver modules	40 000 €	502	20 100 000
LT receiver modules	17 000 €	561	9 500 000
Heat exchanger	3 M€50MW	200 MW	12 000 000
Heat recovery boiler	1 M€ 50MW	200 MW	4 000 000
Pumps & Piping	.4 M€ / 50MW	200 MW	1 600 000
Sum			149 600 000
Safety surcharge	10%		15 000 000
Total Investment I <sub>0</sub>			165 000 000



## **Operational costs**

Fixed costs	Unit price	Units	Total [€a]
Personal	70 000 €a	10	700 000
Insurance	$2\%$ of $I_0$		3 300 000
Maintenance	$4\%$ of $I_0$		6 600 000
Redox System	3.7 M€a for 50 MW	(for 200 MW)	14 800 000
Variable costs			
Separation Unit	2633 k€a for 50 MW	(for 200 MW)	10 532 000
Nitrogen	294 k€a for 50 MW		1 176 000
Water	11 k€a for 50 MW		44 000
Electricity	1777 k€a for 50 MW		7 108 000
Sum			44 260 000



#### **Conclusion and Outlook**

- ▼ Solar fuels have a huge potential to become the energy carriers of the future especially for mobile applications
- → Thermal production is more efficient than electrolysis
- The solar research of DLR will continue its work on reforming and cracking of carbonaceous materials (e.g. natural gas) because of the short and medium term chances of these technologies
- → Solar thermal up-grading of biomass is not efficient.
- Thermochemical cycles for hydrogen production are for long term use
- ▼ Three main problems have to be solved:
  - Heat provision
  - Material properties of components and reactants
  - Product separation
- The aim is to produce renewable fuels by ecologically and economically reasonable technologies



#### **Acknowledgement**

The Projects HYDROSOL, HYDROSOL- II; HYTHEC, SCR, SOLASYS, SOLREF, SOLHYCARB, Hi2H2, and INNOHYP-CA have been co-funded by the European Commission.

#### HYDROSOL has been awarded

- → Eco Tech Award Expo 2005, Tokyo
- → IPHE Technical Achievement Award 2006
- → Descartes Prize 2006





